

Simultaneous Detection of Tricyclic Antidepressants using Higher Performance Liquid Chromatography

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ABSTRACT

Despite the introduction of newer antidepressants, tricyclic antidepressants are still widely used for the treatment of major depression, chronic pain and anxiety. And is the most leading cause of drug-induced death in many patients. Monitoring their blood/plasma concentration is therefore a priority in forensic toxicology. In this study, a quick and rapid method for simultaneous detection and quantification of three TCAs (desipramine, nortriptyline and amitriptyline) in whole blood was developed and validated on HPLC-UV under a gradient elution, using a C₁₈ column and a mobile phase mixture of ACN (35% v/v) and a pH 3.5, 50mM phosphate buffer (65% v/v). The evaluated validation parameters were satisfactorily accepted in accordance with ICH and FDA guidelines. The determination coefficients were more than 0.99 within the tested concentration range (0.1–1mgL⁻¹), LOD and LOQ were between 0.0027-0.0051 and 0.27-0.051 respectively, Accuracy expressed as % Er was found to be between 0.1-7% and precision expressed as %CoV was found to be between 2-10%. All compounds were satisfactorily determined in one single injection within 20 min. sample pre-treatment was optimised on three sets of cartridges (C₂, C₁₈ and Strata-X) on a solid phase extractions, with C₁₈ giving the best recovery for all analytes ranging from 60-105%. The developed method proved to be suitable for routine work and it was used to successfully analyse forensic blood samples.

Key words: Antidepressants, HPLC, tricyclic, toxicology and simultaneous.

INTRODUCTION

Depression is a mental disorder that is characterised by low self-worth, low mood, fatigue and loss of appetite³⁵. Pharmacological treatment has immensely advanced since the development of the tricyclic antidepressants between the period of 1960 and 1980³¹. Although it gained clinical recognition in the 1950s^{12,15} the tricyclic antidepressants (TCAs)

and monoamine oxidase inhibitors were originally adopted. However, when compared with monoamine oxidase inhibitors. The TCAs offer advantages over monoamine oxidase inhibitor (MAOI) antidepressants because they have less drug to drug interaction and with a lower therapeutic index are much more effective in the treatment of depression.

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The TCA's also appear to be reducing the suicidal risk compared to the MAOI's^{10,25}. This therefore prompted the development of newer and safer antidepressants. However, TCA's are still widely used despite the development of new classes of antidepressants, such as the selective serotonin reuptake inhibitors (SSRIs), serotonin–norepinephrine re-uptake inhibitors (SNRIs), norepinephrine–dopamine reuptake inhibitors (NDRIs), and norepinephrine reuptake inhibitors (NRIs)^{17,27} due to their effectiveness and cost^{25,31}. TCAs are also regularly prescribed for treatment of a variety of other disorders including for example anxiety, eating disorders, attention deficit hyperactivity disorder, and enuresis in children. It has also been used as a treatment for neuropathic pain which explains why they are still widely in use^{9,29}.

In a study were TCAs and SSRI were compared in terms of tolerability and potency, TCAs appeared more effective inpatients than SSRI, and however, SSRIs are better tolerated, due to their milder adverse effects than TCAs¹. This may be because TCAs act on numerous neurons, including serotonergic, dopaminergic and adrenergic neurons while SSRIs act specifically on serotonergic neurons, therefore monitoring plasma concentration is not required with SSRIs as in the case of TCAs²⁸.

The prevalence use of TCAs on suicide attempt keep increasing as it remains the leading cause of drug-death in many countries¹³. Between the period of 1993-2004 England and Wales recorded 5,602 deaths caused by antidepressants and this accounted for about 15% of all deaths due to poisonings. Fatality typically occurs with ingestion of 20 mg/kg body weight however it is not possible to relate a dose to a post mortem blood drug concentration therefore the dose even if confirmed is a poor predictor of the resultant effect. This is in part due to the variation in metabolism and ethnicity³⁰ as previously indicated but also due to the phenomenon of post mortem drug redistribution^{16,21}.

The development of High Performance Liquid Chromatography (HPLC) began in the late 1960s and early 1970s, it is a

unique form of chromatography commonly used in analysis, to identify, separate, and quantification of simple and complex compounds. It is a conventional method applied in several areas which include but not limited to forensic toxicology, pharmaceuticals, biotechnology, environmental, polymer and food industries^{3,34}. HPLC provides a means for isolating analytes from potential interfering impurities to allow detection, identification and quantification. This made HPLC more a prevalent technique in the analysis of TCAs either in combination with UV or MS detection since most drugs will breakdown at high temperature therefore gas chromatography (GC) tends not to be used^{11,31}. It remains the most widely used techniques to analyse TCA drugs despite other analytical methods being available¹⁴. Both normal and reversed-phase HPLC (RP-HPLC) methods are used in the analysis of TCAs³¹.

We aimed to develop a method on HPLC-UV for the simultaneous detection of TCA's (Amitriptyline, Nortriptyline and Desipramine) and validate the developed method as well as to optimise and apply the method developed using SPE on whole blood.

MATERIAL AND METHODS

Study Design

This research study take in cognisance with the risk assessment and the ethical approval was granted.

SPE manifold, vacuum pump, SPE cartridges (C2, C18 by Varian and strata X by Phenomenex), Ammonium, Acetonitrile, Methanol, Nitrogen and collecting tubes. 1ml of whole blood spiked with 1mgL⁻¹ of TCAs was transferred to Eppendorf and either centrifuged at 4000rpm for 10mins or sonicated for 15mins depending on the pre-treatment type.

The SPE procedure was evaluated according to the methods described by Bielicka-Daszkiwicz et al⁴., and Camel⁷. This consisted of evaluating the influence of parameters in each of the standard four steps thus, (i) conditioning of the sorbent; (ii) isolation of the analytes; (iii) washing of sorbent (iv) elution step.

Method Application

Three types of SPE cartridges (C2, C18, and Strata-X) were tested. The first two columns belong to the group of reverse-phase sorbent. The Strata-X column is a hydrophilic polymeric sorbents (mixed-mode) bonded silica containing both reversed phased and cation ion exchange sorbent, where the starting methods for SPE methods are similar. However, the results obtained on the C2 cartridge did not allow the use of this column to isolate TCAs from spiked blood samples and this cartridge was discarded. This might be due to fact that C18 contains octadecyl (higher alkyl), while C2 contains ethyl (lower alkyl chains) as reactive site. More so, the lower the alkyl chain the lesser the retention. The remaining two cartridges were further optimised from the starting method, however

during the optimization process, sample pre-treatment, loading conditions, washing reagents, and volume of eluent were considered. Whole blood was selected, as this is mostly encountered in most forensic toxicological analysis.

RESULTS AND DISCUSSION

To improve the recovery of the TCAs, the washing reagents, elution volume/type were changed (See Table 1).

Table1. Methods 1-5, showing SPE optimization steps involving sample pre-treatment, washing reagents and elution solvents. The elution volume were sequentially applied 4 times in the order; 1ml, 0.5ml, 0.5ml and 0.5ml and the eluent were collected 4 times for each cartridge used on each method.

EXPT 1	1	2	3	4	5
STEPS	C18	C18	STRATAX	STRATAX	STRATAX
CONDITIO	1ml CH3OH	1ml CH3OH	1ml CH3OH	1ml CH3OH	1ml CH3OH
	1ml H2O	1ml H2O	1ml H2O	1ml H2O	1ml H2O
LOADING	sonicate	sonicate	centrifuge	sonicate	sonicate
WASHING	1ml H2O	1ml H2O	1ml H2O	1ml H2O	1ml H2O
	1ml CH3CN	1ml CH3OH	1ml CH3OH	1ml CH3OH	1ml CH3CN
	1ml CH3CN	1ml CH3OH	1ml CH3OH	1ml CH3OH	1ml CH3CN
VACDRY (5MINS)					
ELUTION	(CH3CN:NH4OH)	(CH3OH:NH4OH)	(CH3OH:NH4OH)	(CH3CN:NH4OH)	(CH3CN:NH4OH)
	(98:2)	(98:2)	(98:2)	(98:2)	(98:2)
	0.5ml*(2)	0.5ml*(2)	0.5ml*(2)	0.5ml*(2)	0.5ml*(2)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
RECONSTI	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml mobile phase
EVAPORATE TO DRYNESS UNDER NITROGEN AT 40 DEGREE CELCIUS					
1ml of TCAs containing amitriptyline, nortriptyline and desipramine were spiked into 1ml of blood before					
subjecting to either sonication for 15 mins or centrifugation at 4000rpm for 10mins					

Moreover, the first method was not sufficient to yield good result but it was not surprising i.e. eluting with Acetonitrile: Ammonium hydroxide ($\text{CH}_3\text{CN}:\text{NH}_4\text{OH}$) ratio (98:2). Since the retention of basic drugs in aqueous sample on reversed-phase SPE sorbents is not only due to weak hydrophobic interactions between analyte and apolar group of silica modified surface, but also strong ionic interaction between the drug in its charged form and with charged residual silanols¹⁹. This therefore prove that acetonitrile either alone or combination with ammonium hydroxide is not sufficient to break the hydrophobic interaction. This findings corroborated another findings that discover acetonitrile cannot interact with silanols via hydrogen bonding however dipole–dipole interactions are possible⁶. More so, in another study where acetonitrile and methanol were compared as wash solvent for basic drugs, more recovery was obtained when acetonitrile was used as wash solvent compared to when methanol was used³³. Since a good eluent is supposed to be strong enough to elute the compounds of interest in a limited volume irrespective of the binding force, acetonitrile was therefore alternated as a washing solvent. This was however in contrast with the claimed that methanol was unable to elute TCAs on C_{18} ³⁶.

Although, methanol was suggested to be too strong as wash solvent, yet it was used as wash solvent while the eluent solvent was changed (Table 1) to methanol and ammonium hydroxide (NH_4OH) the result of this is shown in Figure 1 (a, b & c). Since the breakthrough volume is a function of the analyte hydrophobicity and the mass of the sorbent used²⁶.

The first elution volume was able to break the van der Waals for amitriptyline whereas additional volume was required to elute both desipramine and nortriptyline. This phenomenon might be attributed to the structure of the analytes. Amitriptyline is a tertiary amine with three carbon substitutes located on the terminal nitrogen of its side chain, which gives it more affinity to the eluent solvent and at the same time easy for disrupting its binding force whereas desipramine and nortriptyline are demethylated metabolite of imipramine and amitriptyline respectively and are secondary amines; with two carbons on the terminal nitrogen of the side chain.

This therefore supports the claim that retention of analyte is a function of the analyte's structure, rather than, on interactions of its functional groups with the sorbent surface³⁷. This procedure is similar to a study where olanzapine (a typical basic analyte) was extracted, methanol was used as wash solvent and methanol/ NH_4OH was also used as elution solvent and a higher recovery was observed¹⁹. Although its recovery was more than the observed recovery on $\text{CH}_3\text{OH}:\text{NH}_4\text{OH}$ (98:2). However, the observed recovery on $\text{CH}_3\text{OH}:\text{NH}_4\text{OH}$ (98:2) was better than that observed on similar column by Madej et al²³.

It is noteworthy to mention that adding a competing cation (NH_4OH) to the organic elution solvent block the charged silanol groups of the C_{18} sorbent thereby allowing maximum elution of the adsorbed analytes¹⁹. See percentage recovery in Figure 1d.

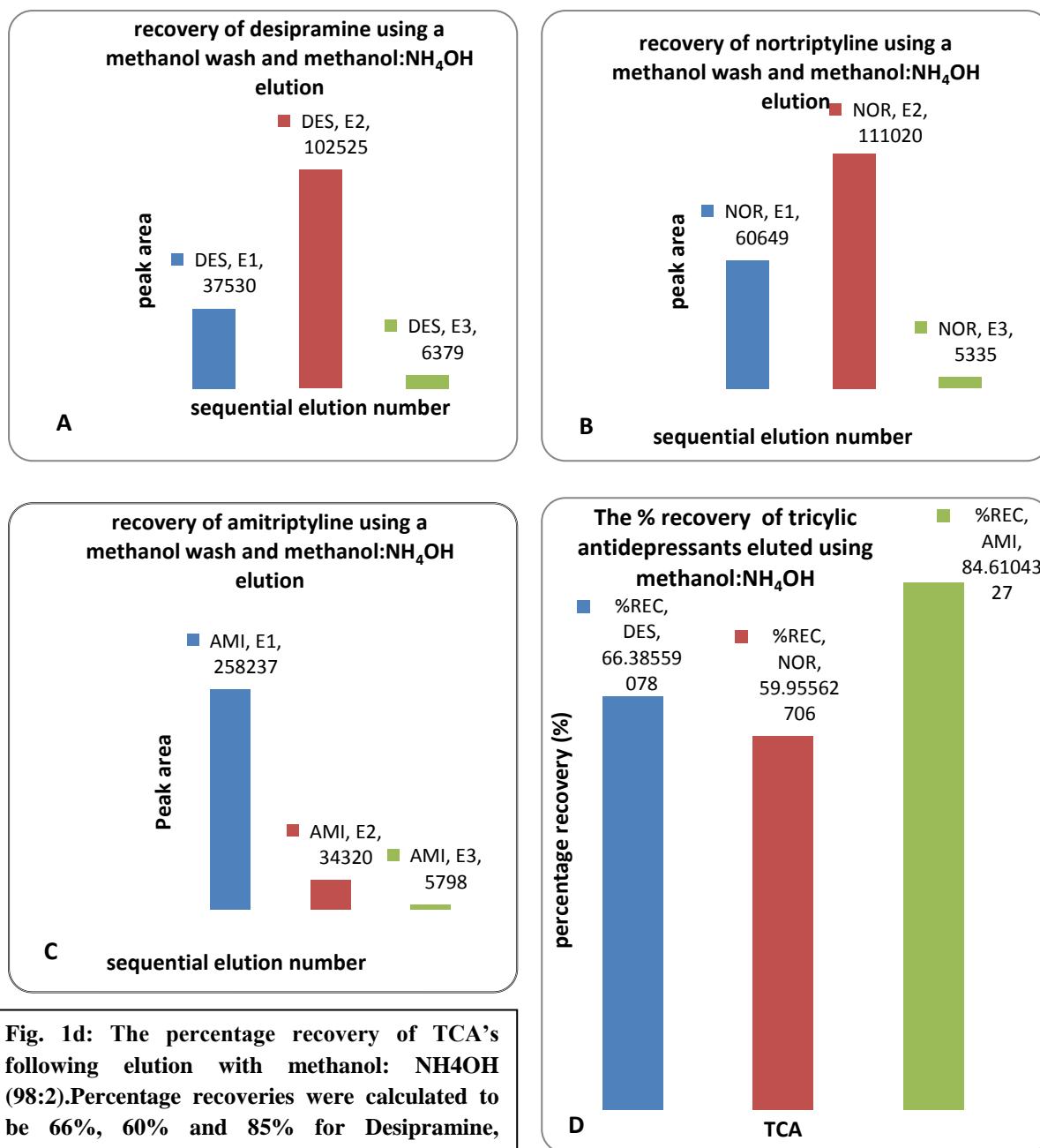


Fig. 1d: The percentage recovery of TCA's following elution with methanol: NH₄OH (98:2). Percentage recoveries were calculated to be 66%, 60% and 85% for Desipramine, Nortriptyline & Amitriptyline respectively

Fig. 1 (a-d): The effect of sequential elution for the recovery of the tricyclic antidepressants (desipramine (a), nortriptyline (b) and amitriptyline (c)).

Note E1 = 1ml, followed by E2=0.5ml and finally E3=0.5ml of methanol: NH₄OH

Furthermore, the effects of sample pre-treatment was assessed between two column types (C18 and Strata X) – methods 2 and 3. Unfortunately, all Strata-X methods (3-5) yielded no good results indicating that the method does not apply to Strata-X **Sample Pre-treatment.**

TCA's are extremely bounded to plasma-proteins, majorly alpha1-glycoprotein,

albumin and lipoproteins. During the analysis of these drugs especially at extraction process, the binding reduce the percentage recovery of the drugs because their active site have been occupied by the protein binding resulting in ineffective interaction between the drugs and the sorbent binding site more so, the larger the protein the more it block the sorbent pores thereby leading to clogging and the drug

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therefore pass through the sorbent un-retained. However in this study disruption of cell membrane and separation of blood components was tested (since clogging was experienced in the starting method) by two techniques including sonication and centrifugation. But sonication was finally chosen as it simultaneously allows moderate flow of the blood and sufficient retention between TCAs and the non-polar sorbent of the cartridge whereas centrifugation flows were too fast as it does not provide enough retention for the drugs to adsorb to the sorbent. Choosing sonication was consistent with previous studies³³, although it was not compared with any other techniques.

A study claimed that reconstitution plays a role in recovery²² thus reconstitution was therefore tried using mobile phase on Strata-X (CH₃CN:NH₄OH) but unfortunately the peaks were very poor suggesting that reconstitution is independent on recovery provided that

reconstitution is done with appropriate polar solvent. Although the method was not effective on Strata-X, thus observing reconstitution effect on Strata-X does not give a conclusive remark on effect of reconstitution.

Optimisation of Strata-X on loading pH, and elution solvents.

Table 2. The SPE procedure for methods 1-6 on the Strata-X column. The methods show changes in pH conditions of the columns before loading the pre-treated samples and elution solvents on each column. The elution volume was increased from 1ml to 1.5ml in the 1st elution and 0.5ml in the subsequent elution for a total of 4 times.

Figure 1: The effect of sequential elution for the recovery of the tricyclic antidepressants (desipramine (a), nortriptyline (b) and amitriptyline (c)). Note E1 = 1ml, followed by E2=0.5ml and finally E3=0.5ml of methanol: NH₄OH.

EXPT 2	1	2	3	4	5	6
STEPS	STRATAX	STRATAX	STRATAX	STRATAX	STRATAX	STRATAX
CONDITIO	1ml CH30H	1ml CH30H	1ml CH30H	1ml CH30H	1ml CH30H	1ml CH30H
	1ml H2O	2ml Buffer	1ml H2O	1ml H2O	2ml Buffer	1ml H2O
LOADING	Sonicate	Sonicate	Sonicate	Sonicate	Sonicate	Sonicate
WASHING	2ml H2O	2ml H2O	2ml H2O	2ml H2O	2ml H2O	2ml H2O
	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN
	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN	1ml CH3CN
VACDRY (5MINS)						
ELUTION	(CH30H:NH40H)	(CH30H:NH40H)	(CH3CN:NH40H)	(ACN:MeOH)	(ACN:MeOH)	(ACN:MeOH)
	(98:2)	(98:2)	(98:2)	(60:40)	(60:40)	(98:4)
	0.5ml*(3)	0.5ml*(3)	0.5ml*(3)	0.5ml*(3)	0.5ml*(3)	0.5ml*(3)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
RECONSTITUTION IN 1ml CH3CN						
1ml of TCAs containing amitriptyline, nortriptyline and desipramine were spiked into 1ml of blood before						
subjecting the sample to sonication for 15 mins						
EVAPORATE TO DRYNESS UNDER NITROGEN AT 40 DEGREE CELCIUS						

Strata-X is a surface-modified styrene skeleton with a pyrrolidone group, whose retention mechanisms exploit hydrogen-bonding, aromatic and hydrophobic interaction with the analytes for the extraction of basic, acidic, amphoteric and neutral analytes²⁶. However, the extraction of basic drugs requires the analyte to be ionized at pH either two

units/more lower than the pK_a of the compounds. At this pH, approximately 99% of the group is charged³⁶. TCAs are expected to be ionised under the experimental conditions. Unfortunately, no appreciable recovery was achieved in all methods but one (method 5) where the sample was buffered with 50mM phosphate (pH 3.5) before loading.

The recovery is shown in Figure 2 (a,b,&c). Although the rationale for this is unclear, but it has previously been noted that Strata-X exploits both reversed-phase and hydrogen bonding for retention, as such retention of analyte increases with increasing hydrophobicity⁵. Thus, the analytes held too tight/strong to the sorbent which might have led to no recovery on other experiments. The

percentages recovery for this experiment is shown in Figure 2 (d). With the recovery of this method, the Strata-X is still not as good as the C₁₈ cartridge (Figure 1d). TCAs were strongly retained on Strata-X, which can be justified by the strong interaction of NH₂ (amine) group with the carboxylic acid moiety present in the sorbent, thus this account for its low recovery of TCAs².

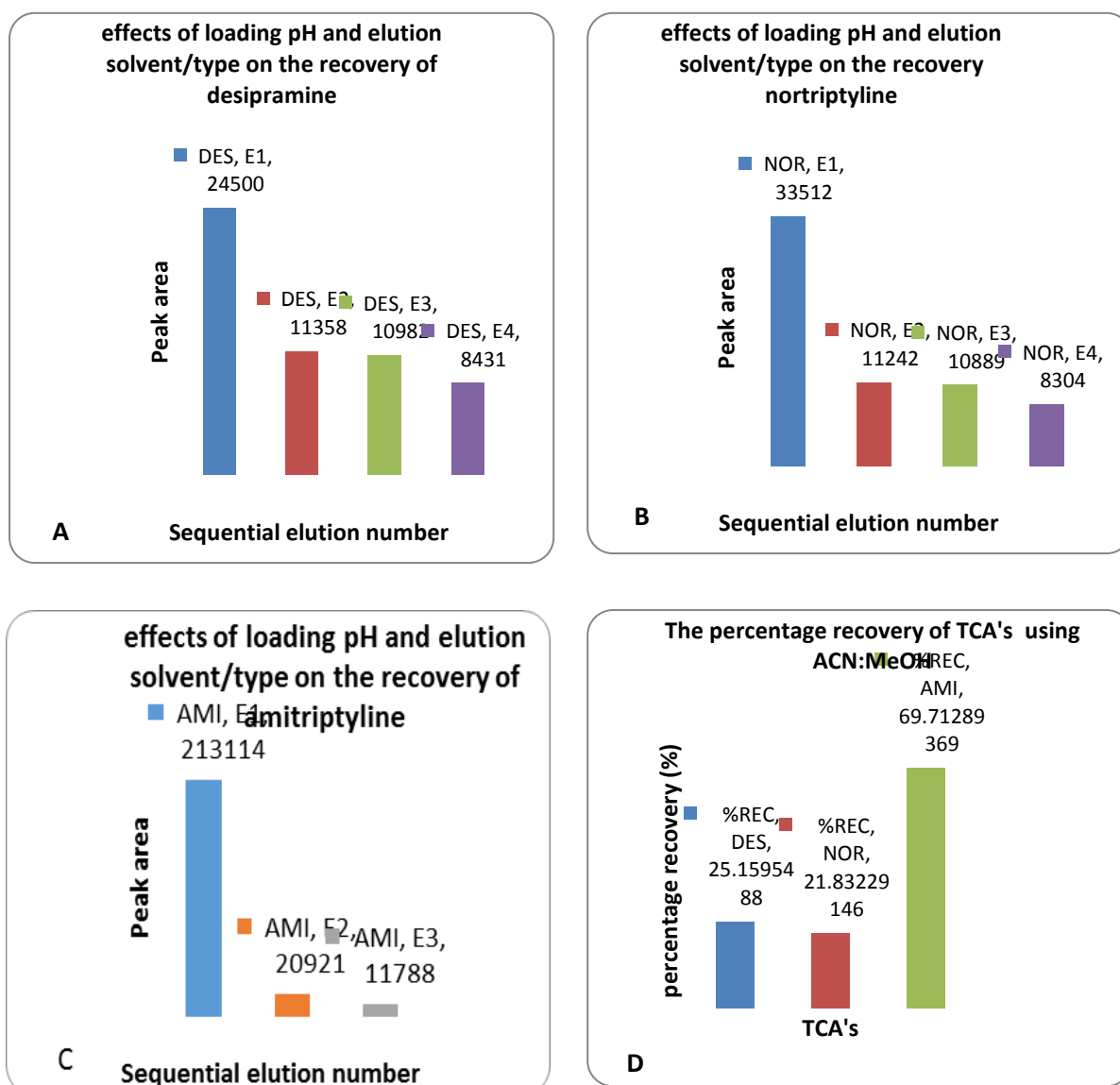


Fig. 2 (d): The percentage recovery of TCAs following elution with Acetonitrile:Methanol in the ratio (60:40). Percentage recoveries were calculated to be 25%, 22%, and 70% for Desipramine, Nortriptyline and Amitriptyline respectively.

Fig. 2: The effect of sequential elution for the recovery of the tricyclic antidepressants (desipramine (a), nortriptyline (b) and amitriptyline (c)). Note E1 = 1.5ml, followed by E2=0.5ml, E3=0.5ml and finally E4=0.5ml of acetonitrile: methanol (60:40).

In each case it can be seen that the 1st 1.5ml elutes the majority of the TCA, but some still remains on the cartridge. Subsequent elution volumes do appear to contain the TCA's as well indicating a larger volume would be required.

Further optimisation of loading pH on Strata-X

It is expected that pH is important to improve recovery; therefore the pH was adjusted using ammonium hydroxide.

Table 3. The SPE procedure for experiment 3, method 1-3, the effect of adding NH₄OH to the elution solvents

EXPT 3	1	2	3
STEPS	STRATAX	STRATAX	STRATAX
CONDITIO	1ml CH30H	1ml CH30H	1ml CH30H
	2ml Buffer	2ml Buffer	2ml Buffer
LOADING	Sonicate	Sonicate	Sonicate
WASHING	2ml H2O	2ml H2O	2ml H2O
	1ml CH3CN	1ml CH3CN	1ml CH3CN
	1ml CH3CN	1ml CH3CN	1ml CH3CN
VACDRY (5MINS)			
ELUTION	(ACN:MeOH:NH40H)	(ACN:MeOH:NH40H)	(CH30H:NH40H)
	(60:40:2)	(60:40:4)	(96:4)
	0.5ml *(3)	0.5ml *(3)	0.5ml *(3)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)	0.5ml*(1)
RECONSTITUTION	IN 1ml CH3CN		
1ml of TCAs containing amitriptyline, nortriptyline and desipramine were spiked into 1ml of blood before			
subjecting the sample to sonication for 15 mins			
EVAPORATE TO DRYNESS UNDER NITROGEN AT 40 DEGREE CELCIUS			

Since the combination of both ACN:MeOH in ratio 60:40 as eluent yielded recovery of all analytes, although at unsatisfactory amount. This might be due to the claim that methanol possess the ability to break hydrogen-bonding whereas acetonitrile only have the ability to break dipole-dipole interaction⁶. Thus ammonium hydroxide was then added to the elution solvent; unfortunately only amitriptyline was recovered ACN:MeOH:NH₄OH ratio (60:40:2), see figure 3(A&B). Increasing the percentage of NH₄OH by 4% did not still improve the recovery of analytes, but did improve the

recovery of amitriptyline ACN:MeOH:NH₄OH ratio (60:40:4) Figure 4 (A&B) nor did the combination of methanol:NH₄OH ratio (94:4). This result was in contrary to the observed result where NH₄OH in methanol was claimed to give a higher recovery when used as an elution solvent in the extraction basic of drugs on Strata-X⁵. This therefore suggested that the neutral polymers (Strata-X) is definitely not a primary extraction mechanism for TCAs under the prescribed conditions and this might probably be due to its aromatic structure which allow interaction with aromatic analytes via π - π interactions²⁰.

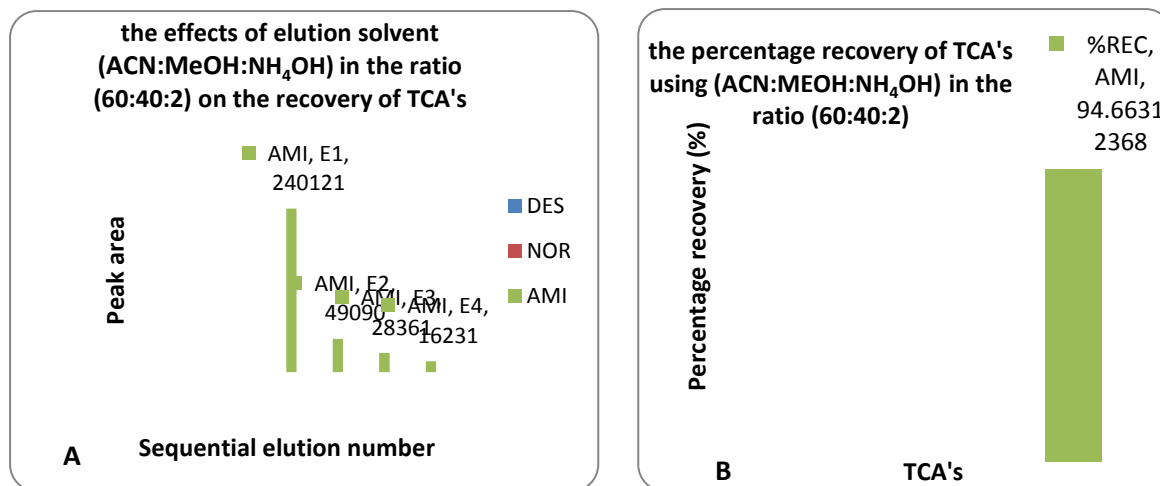


Fig. 3A: The recovery of TCA following sequential elution using ACN:MeOH:NH₄OH in the ratio of (60:40:2). Note E1 = 1.5ml, E2=0.5ml, E3=0.5ml and E4=0.5ml. Amitriptyline was detected in all sequential elution volumes with a percentage recovery calculated to be 95% (Figure 31B). Desipramine and Nortriptyline were not detected.

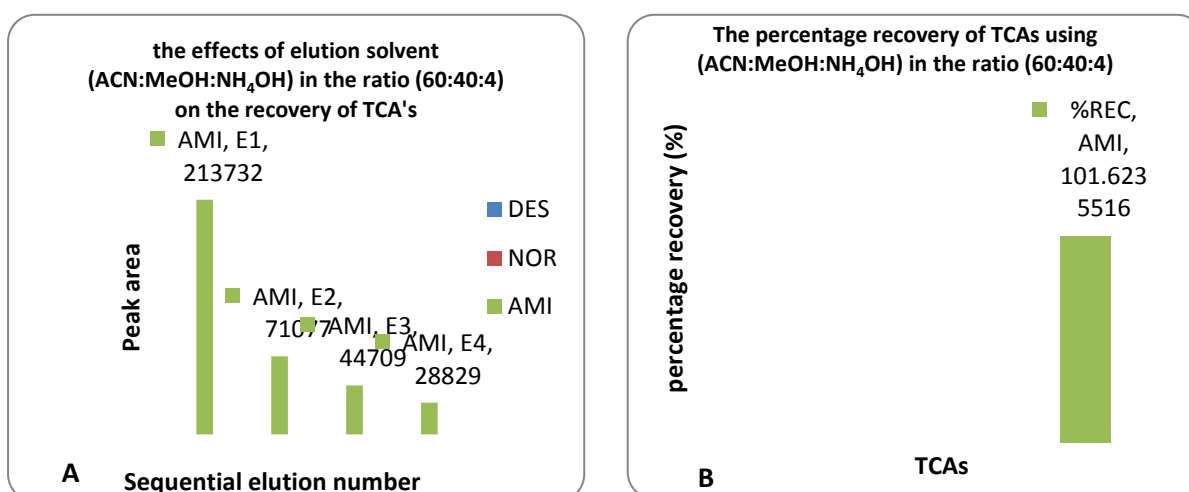


Fig. 4A: The recovery of TCA following sequential elution using ACN:MeOH:NH₄OH in the ratio of (60:40:4). Note E1 = 1.5ml, E2=0.5ml, E3=0.5ml and E4=0.5ml. Amitriptyline was detected in all sequential elution volumes with a percent recovery calculated to be 102% (Figure 4B). Desipramine and Nortriptyline were not detected.

Optimisation of the C₁₈ SPE method

Method 1 in this sequence has already been shown to produce good recoveries of the

TCAs and was used as the starting point for the development of this application method.

Table 4. The SPE procedure for experiment4 method 1-7.Changes to the method includes different wash solvents (method 1&2) and different elution solvents (method3-7)

EXPT 4	M1	M2	M3	M4	M5	M6	M7
STEPS	C18(1)	C18(2)	C18 (3)	C18(4)	C18(5)	C18(6)	C18(7)
CONDITIONING	1ml CH3OH 1ml H2O	1ml CH3OH 1ml H2O	1ml CH3OH 1ml H2O	1ml CH3OH 1ml H2O	1ml CH3OH 1ml H2O	1ml MeOH 1ml H2O	1ml MeOH 1ml H2O
LOADING	Sonicate	Sonicate	Sonicate	Sonicate	Sonicate	Sonicate	Sonicate
WASHING	2ml H2O 1ml CH3OH 1ml CH3OH	2ml H2O 1ml CH3CN 1ml CH3CN	2ml H2O 1ml CH3CN 1ml CH3CN	2ml H2O 1ml CH3CN 1ml CH3CN	2ml H2O 1ml CH3CN 1ml CH3CN	1ml H2O 1ml CH3CN 1ml CH3CN	1ml H2O 1ml CH3CN 1ml CH3CN
ELUTION	(CH3OH:NH4OH) (98:2) 0.5ml*(4) 0.5ml*(1) 0.5ml*(1)	(CH3OH:NH4OH) (98:2) 0.5ml*(4) 0.5ml*(1) 0.5ml*(1)	(CH3OH:NH4OH) (98:2) 0.5ml*(3) 0.5ml*(1) 0.5ml*(1)	(CH3OH:NH4OH) (96:4) 0.5ml*(4) 0.5ml*(1) 0.5ml*(1)	(ACN:CH3OH) (60:40) 0.5ml*(4) 0.5ml*(1) 0.5ml*(1)	(ACN:CH3OH:NH4OH) (60:40:2) 0.5ml *(3) 0.5ml*(1) 0.5ml*(1)	(ACN:CH3OH) (60:40:4) 0.5ml *(3) 0.5ml*(1) 0.5ml*(1)
RECONSTITUTION IN 1ml CH3CN						0.5ml*(1)	0.5ml*(1)
1ml of TCAs containing amitriptyline, nortriptyline and desipramine were spiked into 1ml of blood before subjecting the sample for sonication for 15 mins							
EVAPORATE TO DRYNESS UNDER NITROGEN AT 40 DEGREE CELCIUS							

The elution volume was finally optimised in experiment 4 table17. Thus 2ml of eluent was enough to disrupt any interactive force between the analytes and the sorbent. In addition, the wash solvent was finally optimised between methanol and acetonitrile. Since amitriptyline has proven to possess higher affinity/polarity than other analytes, it was however not properly recovered when methanol was used as wash solvent in figure 5, this suggest that it might have probably been washed away during methanol wash. This findings was also supported in experiment 3 where Strata-X was used, amitriptyline was the only recovered analyte, this probably

suggest that amitriptyline is although structurally similar with other observed TCAs but it mechanism of activities and binding may be however different and more so, it belong to a tertiary group of TCAs whereas the other analytes belong to secondary group. Although the percentage recovery figure 33B was inconsistent with that obtained in figure 2 d, where similar procedure gave a better recovery. It however evident that, it's better to use a less polar solvent such as acetonitrile as washing solvent, if recovery is to be maximised and maintain consistency of results.

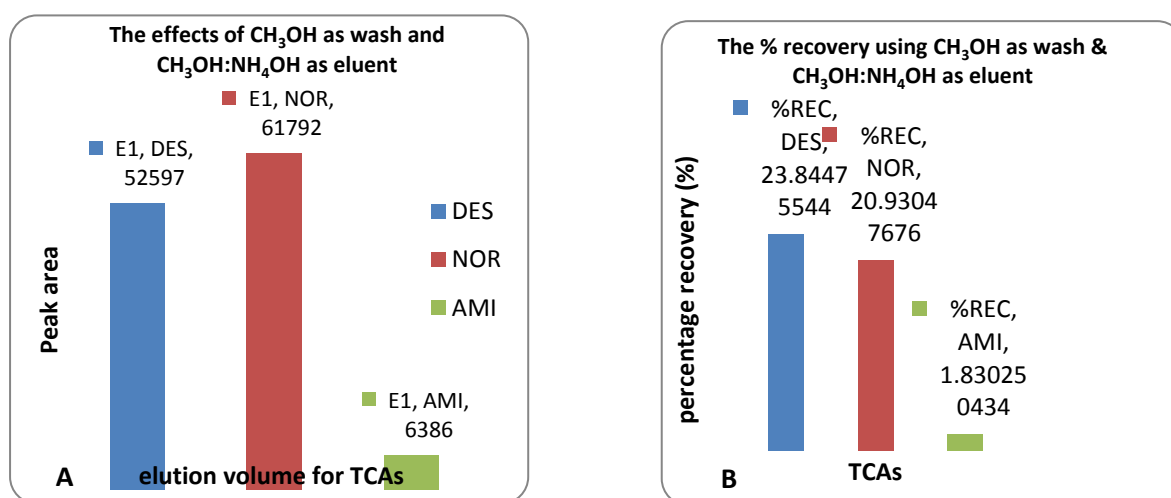


Fig. 5A. The recovery of TCAs following the use of 2ml of (MeOH:NH₄OH) in the ratio of (98:2). E1 signifies that all TCAs were eluted during the 1st elution using 2ml of the elution solvent. Percentage recoveries were calculated to be 24%, 21%, and 2% for Desipramine, Nortriptyline and Amitriptyline respectively (Figure5B).

Using acetonitrile as the wash solvent (method 2,3) but at different elution volumes gave a better a better recovery (Figure 6) when compared with the recovery of both figure 1d

4B where methanol was used as wash solvent. Figure A and B in the ratio (98:2).E1 signifies that all analytes were eluted during the 1st elution using

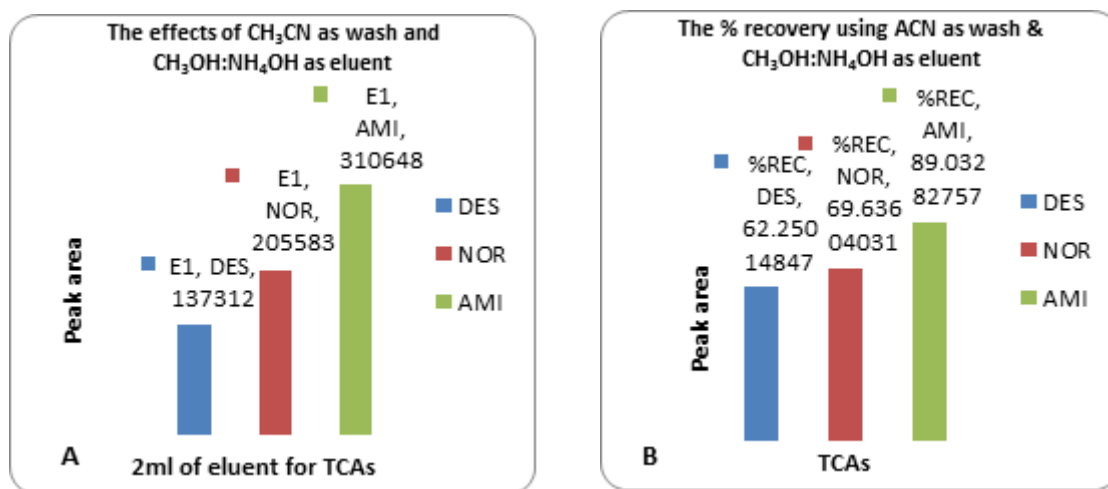


Fig. 5A: The recovery of TCAs following single elution with 2ml of CH₃OH:NH₄OH 2ml of the elution solvent.

The percentage recoveries were calculated to be 62%, 70%, and 89% for Desipramine, Nortriptyline and Amitriptyline respectively (Figure 5B).

Acetonitrile has proven to be a suitable wash solvent in term of consistency in

both methods 2 and 3 Figure. In contrast to methanol wash that resulted into inconsistent recovery in both. Thus acetonitrile was therefore maintained in the subsequent analysis. Figure 6.

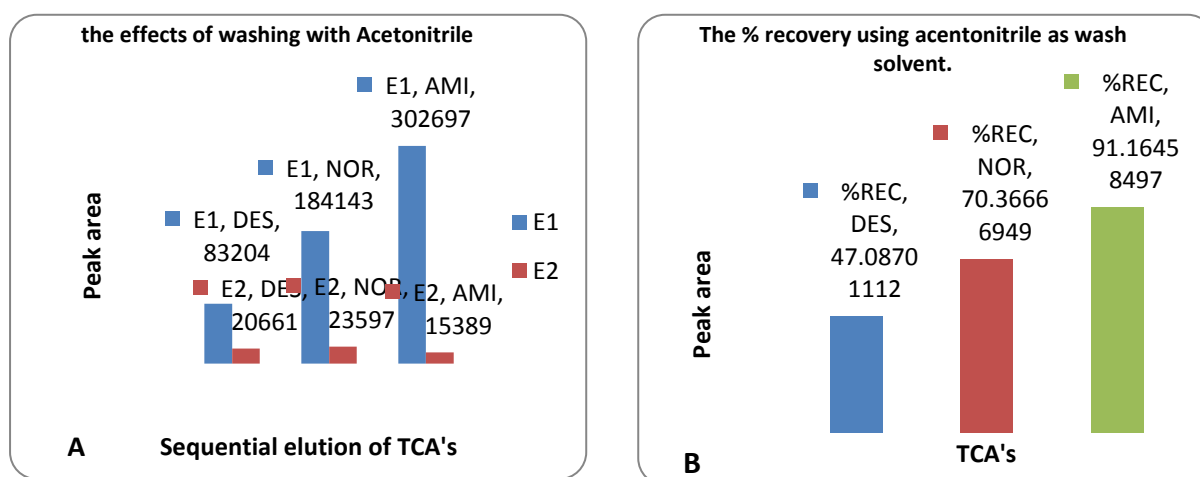


Fig. 6A: The recovery of TCA's following the use of acetonitrile as wash solvent and sequential elution using CH₃OH:NH₄OH in the ratio 98:2. Note E1=1.5ml and E2=0.5ml. The percentage recoveries were calculated to be 47%, 70%, and 91% for desipramine, nortriptyline and amitriptyline respectively (Fig. 6B).

Increasing the percentage of ammonium hydroxide (method 4) although resulted in increased recovery of amitriptyline but

reduced desipramine. But better, compared to the recovery in figure 35b, where 2% NH₄OH was used with 98% CH₃OH as eluent.

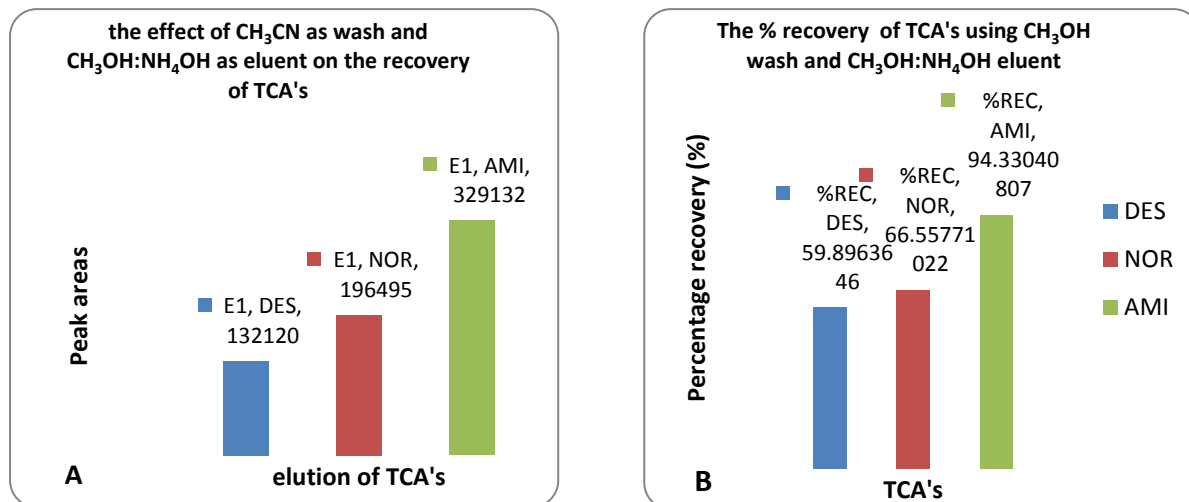


Figure 7A. The recovery of TCAs following single elution with 2ml of CH₃OH:NH₄OH in the ratio (96:4). E1 signifies that all analytes were eluted during the 1st elution using 2ml of the elution solvent. The percentage recoveries were calculated to be 60%, 67%, and 94% for Desipramine, Nortriptyline and Amitriptyline respectively (Figure 7B).

Using a different eluent (method 5), it was expected that a better recovery would be achieved due to the combination of solvent used, however only amitriptyline was

extracted. This was not surprising because the NH₄OH that was claimed to block the charged silanol groups was not in the combination.

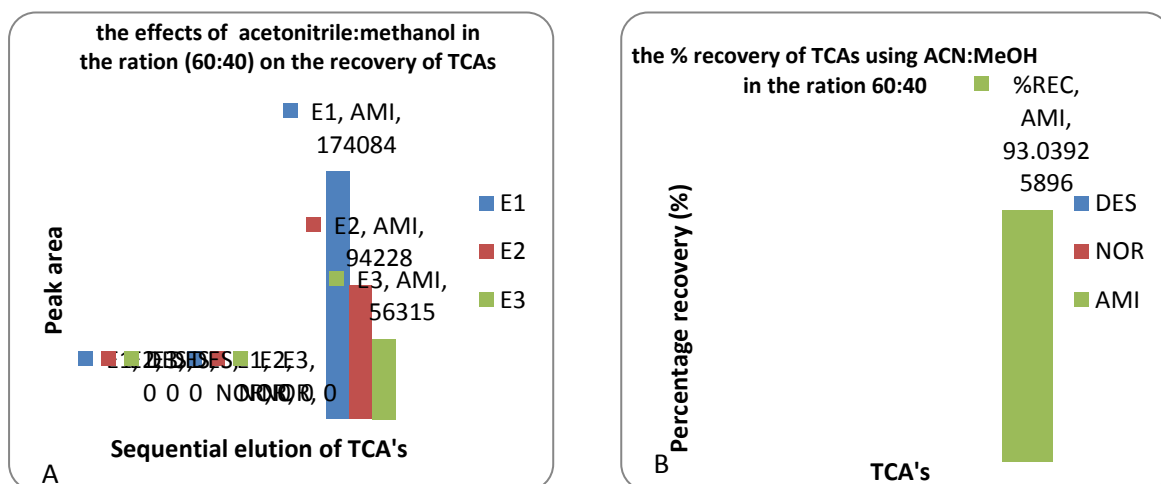


Fig. 8A. The recovery of TCA's following sequential elution using acetonitrile: methanol in the ratio (60:40) E1=2ml, E2=0.5ml and E3=0.5ml. The percentage recovery was calculated to be 93% for Amitriptyline (Fig. 8B). Desipramine and Nortriptyline were not detected.

NH_4OH was therefore added to the subsequent elution solvent (method6) to confirm if

NH_4OH must be used in combination with any intended solvent for elution. Figure 9

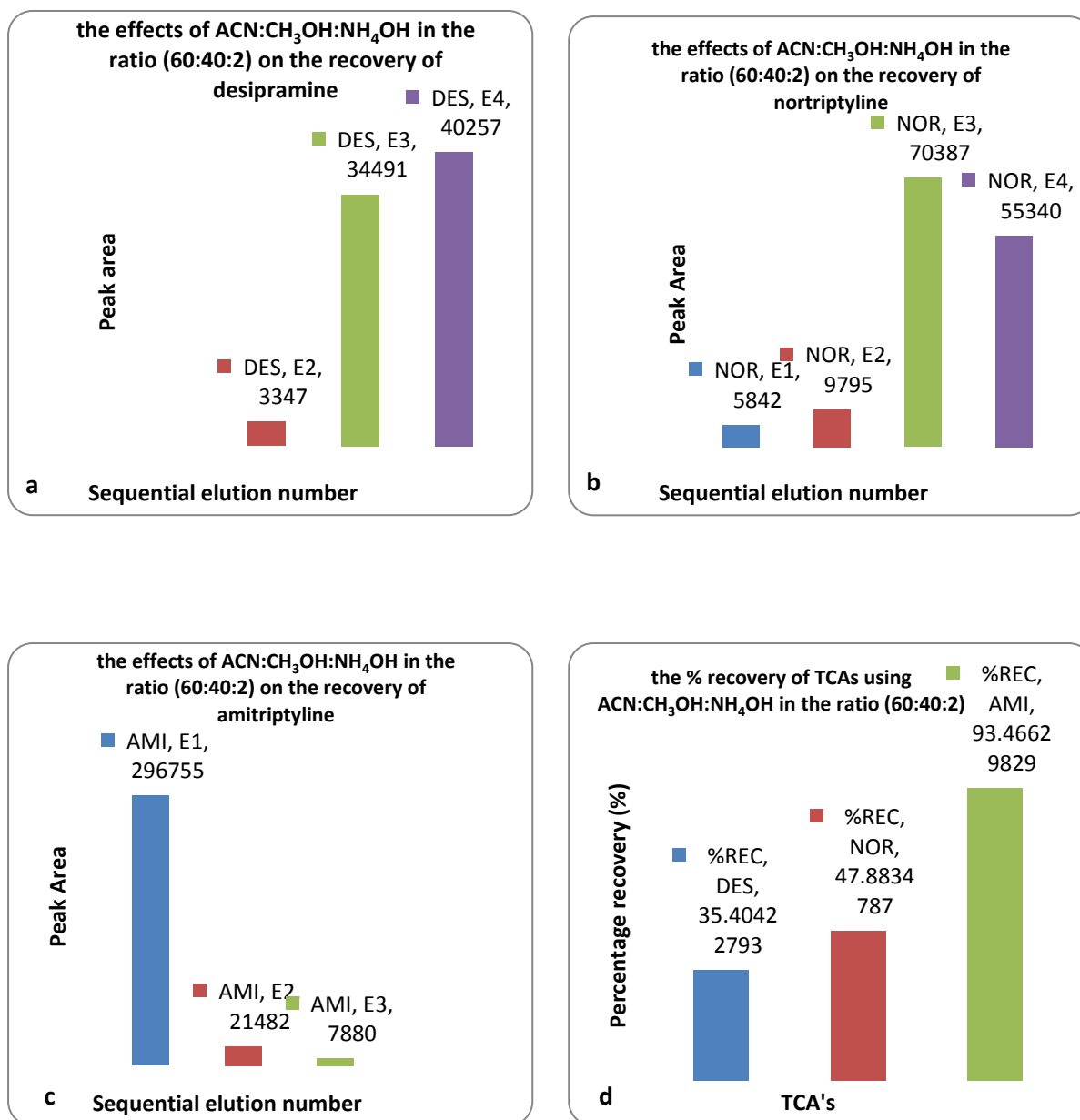


Fig. 9 (a,b&c).The recovery of TCA's following sequential elution using $\text{CH}_3\text{CN}:\text{CH}_3\text{OH}:\text{NH}_4\text{OH}$ in the ratio (60:40:2) E1=1.5ml, E2=0.5ml, E3=0.5ml and E4=0.5ml.The percentage recovery was calculated to be 35%, 48% and 93% for Desipramine, Nortriptyline and Amitriptyline (Fig. 9(d)). The recovery obtained was low but was better when compared with ($\text{CH}_3\text{CN}:\text{CH}_3\text{OH}$) in the ratio 60:40 where desipramine and nortriptyline were not detected/recovered.

The percentage of NH_4OH was further increased) to 4% (method7) and this although

gave a better recovery of TCAs when compared with the 2% NH_4OH .

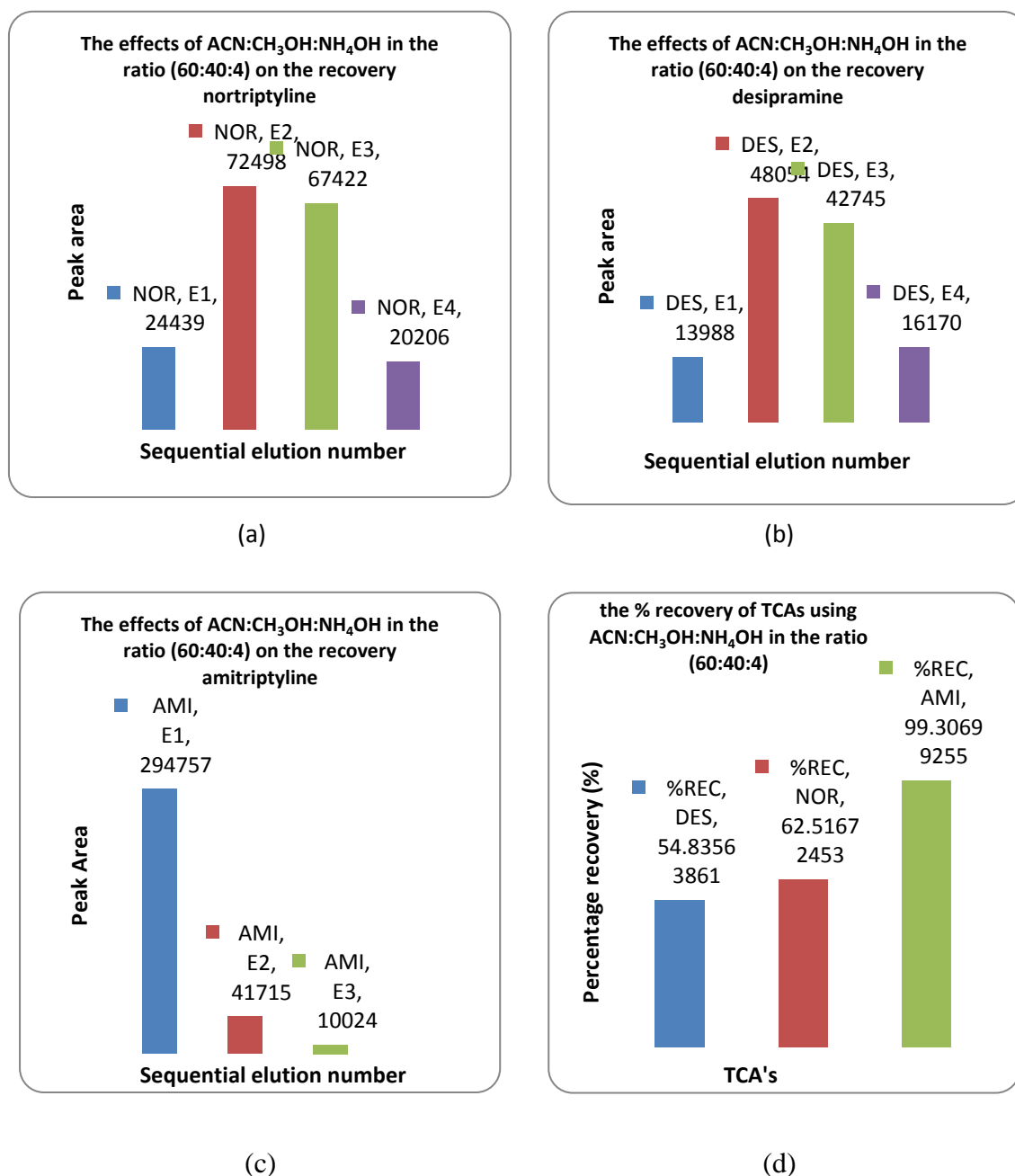


Fig. 10 (a,b&c). The recovery of TCA's following sequential elution using CH₃CN:CH₃OH:NH₄OH in the ratio (60:40:4) E1=1.5ml, E2=0.5ml, E3=0.5ml and E4=0.5ml. The percentage recovery was calculated to be 55%, 63% and 99% for Desipramine, Nortriptyline and Amitriptyline (Fig. 10d).

When the recovery of this method {CH₃CN:CH₃OH:NH₄OH in the ratio (60:40:4)} was compared with that of method4 {(CH₃OH:NH₄OH) ratio 96:4} used as eluent, the CH₃OH:NH₄OH was observed to be better even when both eluent were at the same proportion of NH₄OH (4%). Hence, NH₄OH

must be used with any other polar analyte in the extraction of TCAs. And increasing the level of NH₄OH to at least by (4%) also increases the percentage recovery.

Final optimisation step – influence of loading pH on C₁₈

The final optimisation step was to examine the effects of loading pH of sorbent because in a reversed phase, the pH of the sorbent plays an important role in the retention of the analyte. pH 2 unit below the analyte's pKa create an environment where the analyte is uncharged. Thus, resulting in better retention³⁶. Another

study claimed that addition of water adjust the pH of the sorbent to about 7 and thereby creating a microenvironment for quick and easy⁸. In view of this two, set of pH were achieved by using a 50mM phosphate buffer at pH 3.5 and distilled water (pH about 7).

Table 5: The SPE procedure for experiment 5 method 1&2.

EXPT 5	M1	M2
STEPS	C18(1)	C18(2)
CONDITIONING	1ml MeOH	1ml MeOH
	1ml Buffer	1ml H2O
LOADING	1ml(Sonicated blood	1ml(Sonicated blood+TCA)
WASHING	1ml H2O	1ml H2O
	1ml CH3CN	1ml CH3CN
	1ml CH3CN	1ml CH3CN
VACDRY (5MINS)		
ELUTION	(CH30H:NH40H)	(CH30H:NH40H)
	(96:4)	(96:4)
	0.5ml *(3)	0.5ml *(3)
	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)
	0.5ml*(1)	0.5ml*(1)
EVAPORATE TO DRYNESS UNDER NITROGEN AT 40 DEGREE CELCIUS		
RECONSTITUTION IN 1ml CH 1ml CH3CN		

The effects of loading pH. Since acidic pH was claimed to be better in the extraction of TCAs, method 1 was then buffered with 50mM phosphate buffer at pH 3.5 compared to method 2 which remained loaded with water. The recoveries of TCA's on both methods

were compared in figure 40 and 41. These two methods display almost similar/consistent recoveries and are therefore accepted among all other observed procedures.

Method 1 buffered with 50mM phosphate at pH 3.5

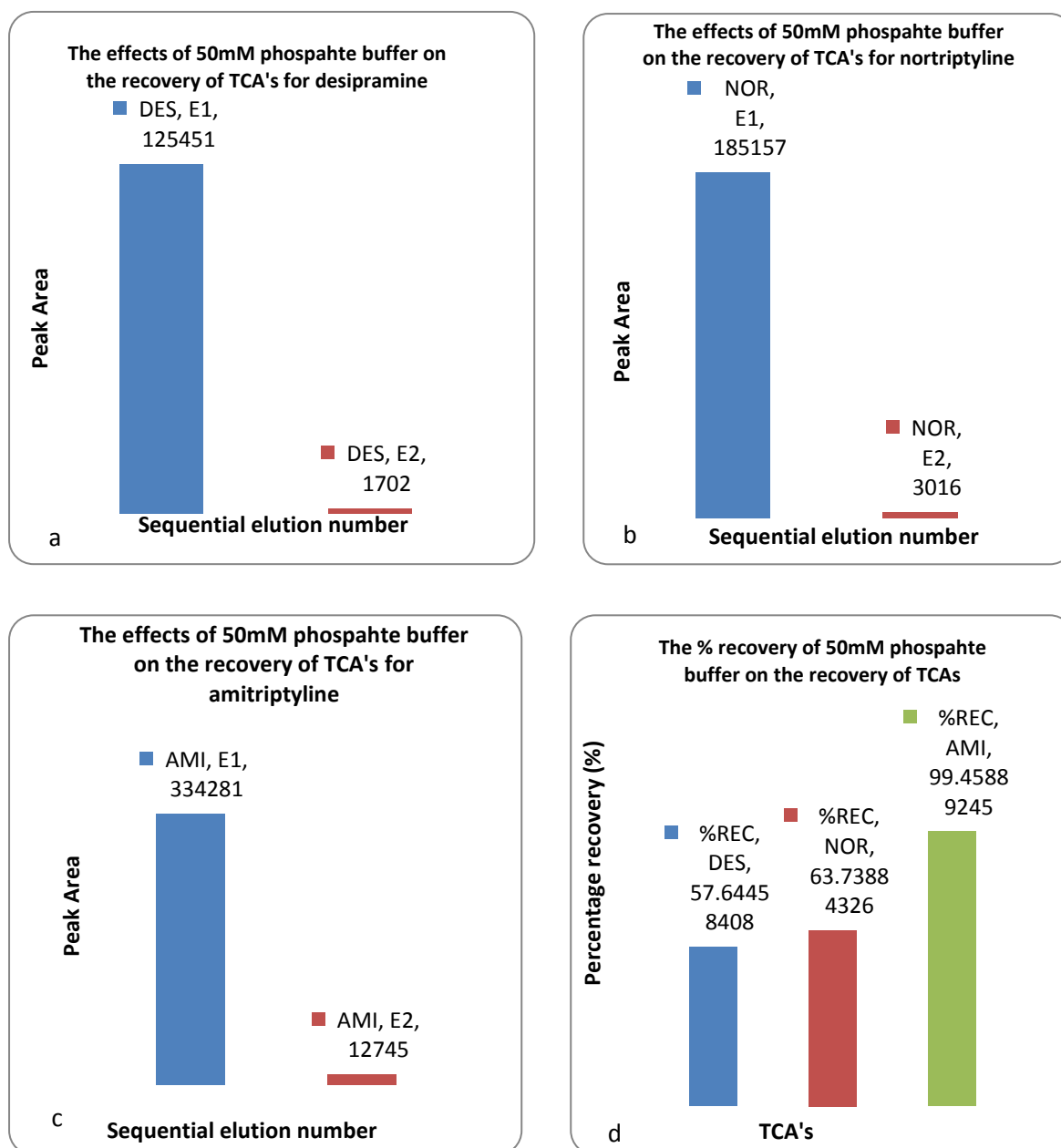


Fig. 11(a,b&c) The recovery of TCA's following loading with phosphate buffer and sequential elution using $\text{CH}_3\text{OH}:\text{NH}_4\text{OH}$ in the ratio (96:4) E1=1.5ml, E2=0.5ml, E3=0.5ml and E4=0.5ml. The percentage recovery was calculated to be 58%, 64% and 100% for Desipramine, Nortriptyline and Amitriptyline respectively (Fig. 40d).

Method 2: buffered with distilled water

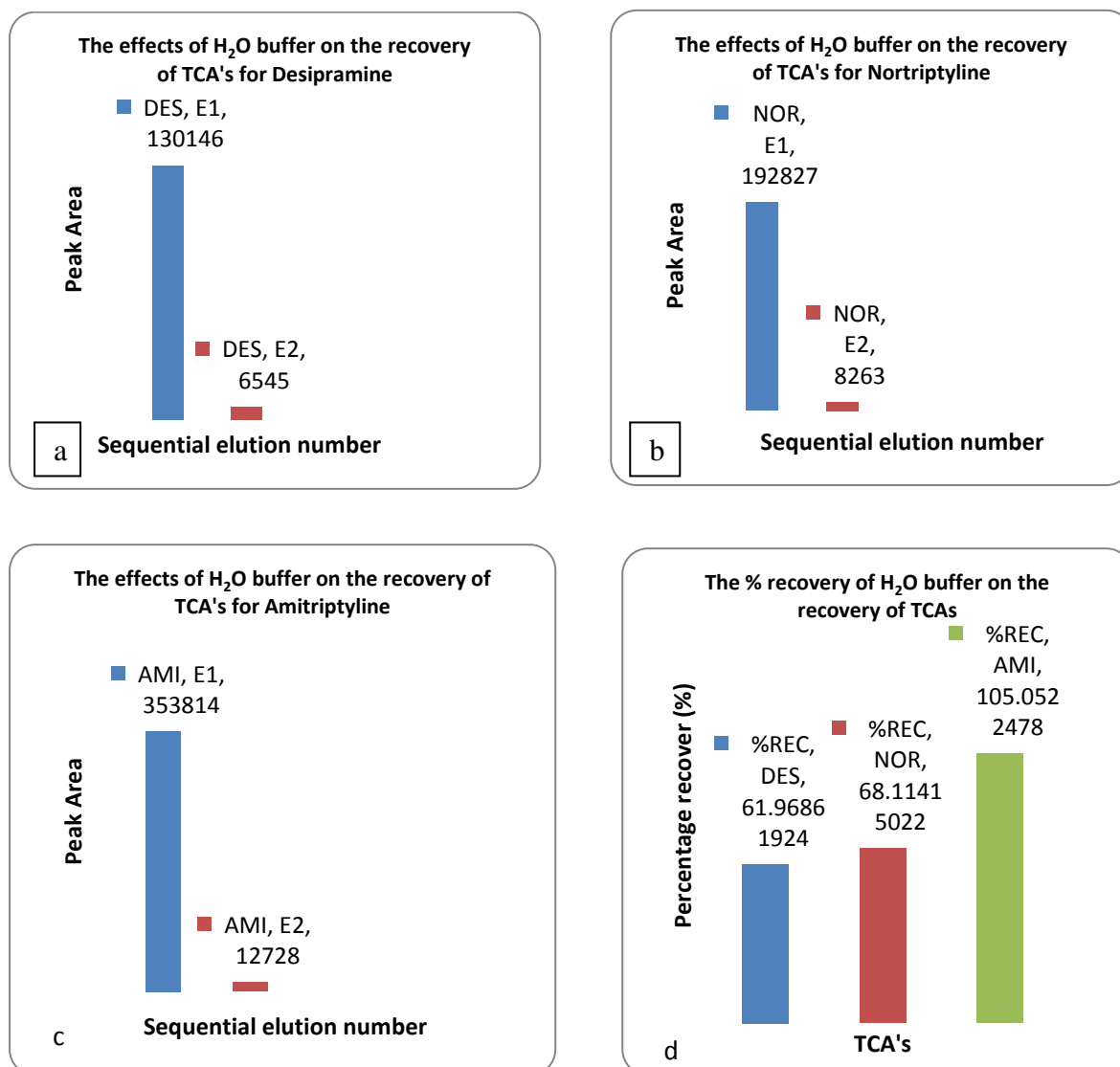


Fig. 12 (a,b&c). The recovery of TCA's following loading distil water and sequential elution using CH₃OH:NH₄OH in the ratio (96:4) E1=1.5ml, E2=0.5ml, E3=0.5ml and E4=0.5ml. The percentage recovery was calculated to be 62%, 68% and 105% for Desipramine, Nortriptyline and Amitriptyline. Of course its recoveries have always been consistent but a little bit change occurs when SPE was performed on different days.

The recovery of amitriptyline in this study was comparable to that observed by Jasinska and Starczewska¹⁸. In addition the recoveries obtained for this method have shown to be better than those obtained by Martinez et al²⁴, and Sanchez de la Torre et al³².

CONCLUSION

A quick and rapid method for simultaneous detection and quantification of three TCAs

(desipramine, nortriptyline and amitriptyline) was developed on HPLC-UV under a gradient elution, using a C18 column and a mobile phase mixture of ACN (35% v/v) and a pH 3.5, 50mM phosphate buffer (65% v/v). The method was successfully validated and the validation parameters were suitably in accordance with ICH and FDA requirement. The validated method was successfully applied

to the analysis and extraction of whole blood with spiked TCAs using SPE technique.

This C18 method gave a satisfactory recovery when compared with all other methods and other cartridges including C2 and Strata-X under the experimental procedures. It has previously been reported using C18 and CH, for the extraction of amitriptyline, imipramine and Chlorprothixene, C18 was claimed to give the best result for amitriptyline (99.5%). It's noteworthy to state one of the significant finding in this research that slight structural differences occur among the TCAs which might account for it binding interactions, retention and recovery.

Limitations of the study

As a result of limited time for the research, some experimental procedures were not subjected to repeats which led to poor reproducibility and consistency of results and finally SPE would be done sequentially rather than simultaneously.

Further work

The eluent pH would be changed since pH of the elution solvent plays a role in the disruption of interactive force, using an acidic methanol such as (98% MeOH:2% HCl) to neutralize the silanol group rather than the basic methanol and other stronger solvents such as methylene chloride, hexane, or ethyl acetate might be tried.

It would be useful to assess other types of cartridges such as the weak cations and hydrophobic due to its ability to increase the retention of basic compounds by strong cationic exchange through its deprotonated carboxylic acid moieties which might therefore increase recovery.

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